

ASYMMETRIC MICHAEL ADDITION OF CHIRAL IMINES TO PHENYLVINYLSULFONE :  
PREPARATION OF KEY CHIRAL BUILDING BLOCKS FOR THE SYNTHESIS OF  
ASPIDOSPERMA AND HUNTERIA ALKALOIDS.

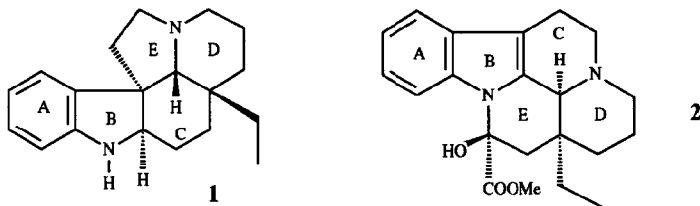
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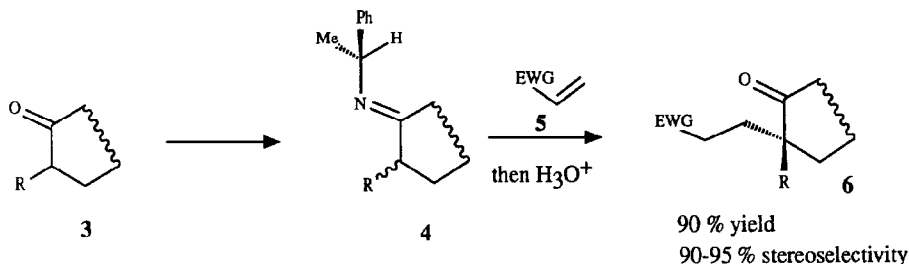
**Abstract :** Addition of chiral imine **8** to phenylvinylsulfone **9** led, after hydrolytic work-up to adduct **11** (91 % stereoselectivity). This derivative has been converted into target cyclopentanones **14** and **16**.

Presence of an ethyl substituent at an angular quaternary carbon center is a common structural feature among the naturally occurring *Aspidosperma* and *Hunteria* alkaloids <sup>1</sup>. It is worthy of note that both opposite configurations at this quaternary center are found in the above indole alkaloids, as exemplified by (+)-aspidospermidine **1** and (+)-vincamine **2** <sup>2</sup>.



Several efficient enantioselective approaches to these molecules have been recently developed, using appropriately substituted butyrolactones or valerolactones as key chiral segments, precursors of the future ring D <sup>3</sup>. In this paper we report on a short and efficient asymmetric synthesis of cyclopentanones **14** and **16** which possess, similarly to the foregoing lactones, the skeleton as well as the functional groups required for a concise approach to alkaloids **1**, **2** and related compounds.

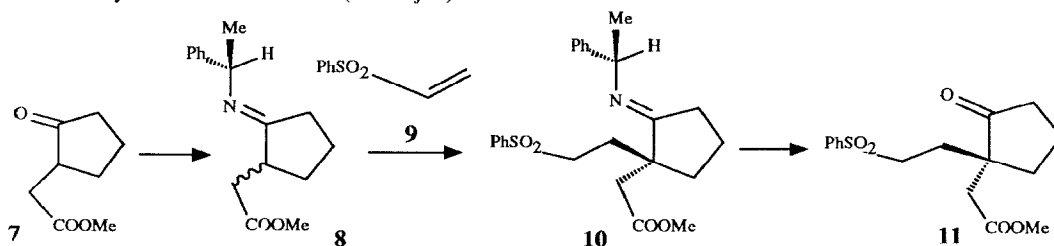
Cyclopentanones **14** and **16** bear a single stereogenic center, namely a quaternary carbon atom in the  $\alpha$ -position to the carbonyl function of the ring. This particular molecular arrangement is encountered in adducts **6** which result from the very efficient asymmetric Michael process which we have disclosed, typified by equation [3  $\rightarrow$  **6**] <sup>4</sup>. Thus chiral imines **4**, derived from *racemic*  $\alpha$ -substituted cyclanones **3** and optically active 1-phenylethylamine, react with electron-deficient alkene **5** to lead, after hydrolytic work-up, to  $\alpha,\alpha$ -disubstituted cyclanones **6** with an excellent yield and a high degree of regio- and enantioselectivity.



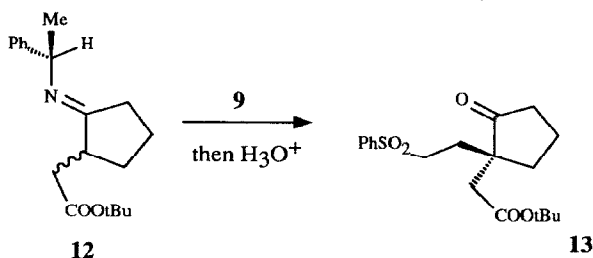
Two geminal appendages -an acetate and an ethyl group- are borne on the quaternary carbon

center of target cyclopentanones **14** and **16**. In the present strategy, ketone **7**, which contains the requisite acetate side-chain, is used as starting material, the ethyl group being next introduced, in an indirect manner, through the aforementioned Michael reaction. For this purpose the corresponding chiral imine **8** was added to phenylvinylsulfone **9**, and the ethyl group, "masked" in the resulting adduct **11**, was then generated by reductive cleavage of the phenylsulfonyl function [**11** → **14** or **16**].

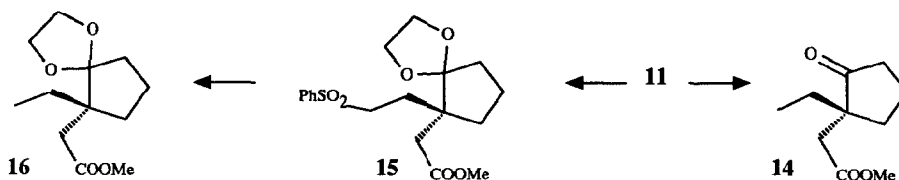
Chiral imine **8** was obtained quantitatively, as nearly equivalent mixture of diastereoisomers, from keto-ester **7** and *R*(+)-1-phenylethylamine ( $[\alpha]_D^{20} + 39.1$ , neat, 96 % ee), (powdered 4 Å molecular sieves, cyclohexane, 20 °C, 72 h). This crude imine was added to phenylvinylsulfone **9** (toluene, 80 °C, 24 h) to lead to imine **10**<sup>5</sup>. The selectivity of the reaction (91 % de) could be determined at this stage by <sup>1</sup>H NMR on an analytic sample of this imine (purified by flash chromatography on silica gel, eluent : hexane/AcOEt : 4/1 in the presence of a few amount of Et<sub>3</sub>N). Hydrolysis of crude imine **10** (10 % acetic acid in water, 36 h at 20 °C) gave (*S*)-compound **11**<sup>6</sup> (74 % overall yield from **7**, 86 % ee, 91 % stereoselectivity). The ee of adduct **11** was established by <sup>1</sup>H NMR, using Eu(hfc)<sub>3</sub> as chiral shift reagent, and its absolute configuration was determined by chemical correlation (*vide infra*).



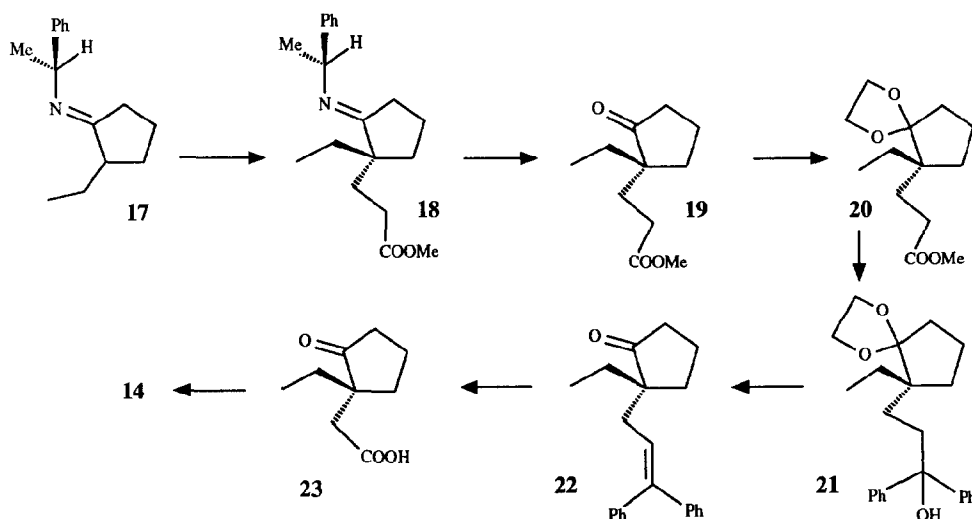
Similarly, addition of imine **12** to phenylvinylsulfone (toluene, 80 °C, 24 h) led, after hydrolytic work-up, to (*S*)-adduct **13**<sup>7</sup> (55 % yield, 86 % ee, 90 % stereoselectivity)



With the desired cyclopentanone **11** in hand, we turned next to its transformation into target compound **14** by reductive cleavage of the phenylsulfonyl function. Cleavage of a phenylsulfonyl group borne on a "non-activated" carbon center requires usually drastic reduction conditions. In the present case this conversion was performed by using a large excess of 6 % sodium amalgam in dry methanol<sup>8</sup>, followed by acidification at pH 4 with acetic acid. The desired (*R*)-keto-ester **14**<sup>9</sup> was thus obtained, though with a low yield (30-35 %). Having attributed this low yield to the presence of the highly electrophilic ketonic function in starting material **11**, this carbonyl was protected by ketalization [**11** → **15**], (10 eq of ethylene glycol, *p* TsOH in refluxing toluene with azeotropic removal of water, 24 h, 90 % yield). As expected, cleavage of the phenylsulfonyl group in ketal **15** proceeds efficiently since, when submitted to the operating conditions of reduction of parent compound **11**, the (*R*)-ketal-ester **16**<sup>10</sup> was produced in a satisfactory yield (65-70 %).



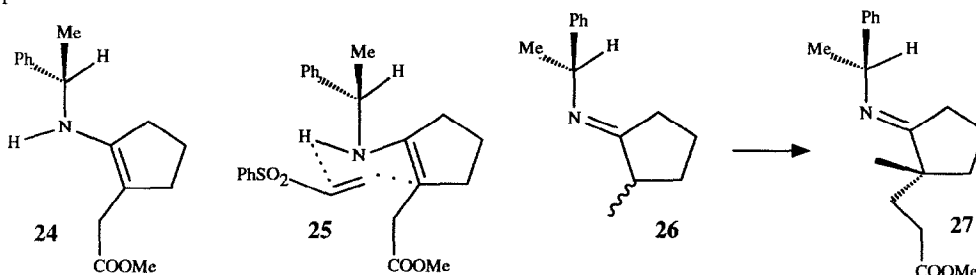
The *R* configuration in target compound **14** was established by the following chemical correlation, using the Barbier-Wieland one-carbon degradation of (*R*)-keto-ester **19**. Imine **17**, prepared from 2-ethylcyclopentanone and *S*-(-)-1-phenylethylamine ( $[\alpha]_D^{20} -40.1$ , neat, 98.5 % ee) was added to methyl acrylate (neat, 40 °C, 24 h), leading to adduct **18** (91 % de by capillary VPC). This adduct was hydrolyzed (10 % acetic acid in water) and the ketonic group in the resulting (*R*)-keto-ester **19**<sup>11</sup> was protected by ketalization (ethylene glycol, *p*-TsOH, 24 h in refluxing benzene with azeotropic removal of water). Ketal-ester **20** was converted into alcohol **21** (excess PhLi, THF, 0 °C) which was transformed into olefin **22** (refluxing acetic acid, 12 h). This olefin was next oxidized into acid **23** ( $\text{RuO}_4$  cat,  $\text{NaIO}_4$ ,  $\text{CCl}_4/\text{H}_2\text{O}$ , 45 °C, 12 h) which was finally methylated by  $\text{CH}_2\text{N}_2$  into desired (*R*)-keto-ester **14**.



## Discussion

It is clear that the nucleophilic species implied in addition [**8**  $\rightarrow$  **10**] is the secondary enamine **24**, in tautomeric equilibrium with imine **8**. The related cyclic, chair-like transition state **25** should be invoked in this addition, in which the proton borne by the nitrogen atom of enamine **24** is transferred to the  $\alpha$ -carbon center of acceptor **9**, *concertedly* with the creation of the new C-C bond ("aza-ene-synthesis-like" transition state)<sup>4b</sup>. The present stereochemical outcome can be easily rationalized, assuming that the alkylation process takes place mainly on the  $\pi$ -face of enamine **24** opposite to the phenyl ring of the chiral amine moiety, when this depicted in its energetically preferred conformation (C-H nearly eclipsing the five-membered ring)<sup>4c</sup>. In other respects, one should note that, somewhat surprisingly, *an increasing of the reaction temperature has no appreciable effect on the stereoselectivity of these Michael additions*; thus, though performed respectively at

20 °C, 40 °C and 80 °C, the three closely related reactions [26 → 27]<sup>4a</sup>, [17 → 18] and [8 → 10] exhibit the same order of diastereoselectivity (90-92 % de). Likewise, presence of a bulky *t*-butyl acetate substituent in imine 12 does not modify significantly the stereoselectivity, when compared with the parent methyl ester compound 8.



The aforementioned Michael additions use 1-phenylethylamine as chiral inductor ; seeing that both isomers of this amine are readily available, inexpensive chemicals (which, in addition, can be easily and quantitatively recovered without any loss of optical activity), *the two enantiomers of target cyclopentanones, 14 and 16 are therefore of comparable accessibility.* Conversion of these chiral building blocks into *Aspidosperma* and *Hunteria* alkaloids is currently under investigation<sup>12</sup>.

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#### References and Notes

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- 10** : pair of diastereoisomers in the 21:1 ratio, determined (250 MHz <sup>1</sup>H NMR) by integration of the doublets corresponding to the methyl group of the amine moiety ; IR (neat) 1675, 1730 cm<sup>-1</sup> ; <sup>13</sup>C NMR (63 MHz, CDCl<sub>3</sub>) major isomer, δ 20.2 24.5 26.6 27.9 28.5 35.4 40.0 46.4 50.8 51.3 61.2 126.1 127.6 127.8 129.0 133.2 139.1 145.1 171.3 177.7.
- 11** : white solid ; mp 73-74 °C (ether) ; [α]<sub>D</sub><sup>20</sup> -1.0 (c = 5, CH<sub>3</sub>OH) ; IR (KBr) 1730 cm<sup>-1</sup> ; <sup>13</sup>C NMR (63 MHz, CDCl<sub>3</sub>) δ 18.2 27.2 33.7 37.0 38.4 47.8 51.1 51.4 127.7 129.1 133.6 138.7 170.7 219.2.
- 13** : solid ; mp 72-73 °C (ether) ; [α]<sub>D</sub><sup>20</sup> -0.6 (c = 5, MeOH) ; IR (KBr) 1730 cm<sup>-1</sup>.
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- 14** : oil ; [α]<sub>D</sub><sup>20</sup> -2.3 (c = 6.2, CCl<sub>4</sub>) ; IR (neat) 1730 cm<sup>-1</sup> ; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) δ 0.86 (t, J = 7.5 Hz, 3H) 1.4-1.5 (m, 2H) 1.8-2.1 (m, 4H) 2.3-2.5 (m, 2H) 2.48 (d, J = 16.4 Hz, 1H) 2.66 (d, J = 16.4 Hz, 1H) 3.63 (s, 3H) ; <sup>13</sup>C NMR (20 MHz, CDCl<sub>3</sub>) δ 8.4 18.7 28.6 32.3 37.6 39.3 49.8 51.5 172.0 221.5.
- 16** : oil ; [α]<sub>D</sub><sup>20</sup> -1.9 (c = 10.5, CCl<sub>4</sub>) ; IR (neat) 1737 cm<sup>-1</sup> ; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) δ 0.90 (t, J = 7.5 Hz, 3H) 1.45 (m, 2H) 1.6-1.8 (m, 6H) 2.31 (d, J = 14.3 Hz, 1H) 2.41 (d, J = 14.3 Hz, 1H) 3.62 (s, 3H) 3.8 (m, 4H) ; MS (m/e) 228 (M<sup>+</sup>) 197 185 169 155 139 127 113 99 86.
- 19** (90 % ee) : oil ; [α]<sub>D</sub><sup>20</sup> -8.4 (c = 3, EtOH). The R configuration was attributed to this keto-ester, assuming that the stereochemical courses in reactions [17 → 18] and [26 → 27]<sup>4a</sup> are the same.
- For a related approach to *Aspidosperma* alkaloids, see ref 4d, 4e.